

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- anode
- cathode
- anion
- cation
- electron movement flow
- direction of ion movement
- half cell reduction equation
- Zn
- Write overall cell equation with phases.  $\text{Zn(s)} + 2\text{Ag}^+(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + 2\text{Ag(s)}$  (10 marks)

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# Voltaic Cells Study Guide

**Donald W. Shive**



## Voltaic Cells Study Guide:

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  - Chapter 2: Essential Elements of Voltaic Cells Study Guide
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