

Worksheet #1
Chapter 12

Q1

How do ethylene oxide molecules interact with each other?



Q2

Which pairs of compounds will form homogenous solutions? For each pair that forms homogenous solutions, what are the intermolecular forces?

- (a) CCl_4 and H_2O
- (b) KCl and H_2O
- (c) $\text{CH}_3\text{CH}_2\text{OH}$ and H_2O
- (d) Br_2 and CCl_4

Q3

- (a) For the pair of compounds $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$ or CH_3OH the one with the higher vapor pressure at a given temperature is _____?
- (b) For the pair of compounds CH_4 or CH_3Cl the one with the higher vapor pressure at a given temperature is _____?
- (c) For the pair of compounds CH_3OH or H_2CO the one with the higher vapor pressure at a given temperature is _____?

Q4

Arrange the following in order of increasing boiling point: $\text{CH}_3\text{CH}_2\text{CH}_3$, $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$, CH_2CH_2 , $\text{CH}_3\text{CH}_2\text{CH}_2\text{Cl}$.

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