

CHEM 101 Worksheet, Chapter 3: Mass Relationships in Chemical Reactions

Multiple choice questions: circle the correct answer(A-E)

Show your work questions (p. 2): show all work! Use the amu for each element from the periodic table!

$$N_{AVO} = 6.02 \times 10^{23}$$

- Atom X has two isotopes: 25.0 wt% is ^{37}X and 75.0% is ^{36}X . What is the average molar mass of X?
 A. 36.0 amu B. 72.0 amu C. 36.5 amu D. 35.5 amu E. 34.5 amu
- The molar mass (in amu) of $\text{ZnSO}_4 \cdot 6\text{H}_2\text{O}$ =
 A. 161.5 B. 179.5 C. 269.5 D. 53.5 E. 215.5
- How many atoms of Na in 5.10 moles of Na_2CO_3 ?
 A. 3.07×10^{23} B. 6.14×10^{24} C. 6.14×10^{23} D. 1.84×10^{25} E. 10.2
- What is the mass in grams of 1.00×10^{12} atoms of lead, Pb?
 A. $3.44 \times 10^{-10}\text{g}$ B. $2.07 \times 10^{114}\text{g}$ C. $1.25 \times 10^{38}\text{g}$ D. $3.44 \times 10^{-34}\text{g}$ E. $1.25 \times 10^{14}\text{g}$
- 0.2 mol Na_2CO_3 = g; 4.24 g Na_2CO_3 = moles
 A. 530g, 449 mol B. 21.2g, 449 mol C. 530g, 0.051 mol D. 21.2g, 0.040 mol E. 16.6g, 0.051 mol
- The wt% composition in tin(IV)oxide is %Sn %O (Sn = tin, element # 50)
 A. 21%Sn, 79%O B. 88%Sn, 12%O C. 12%Sn, 88%O D. 94%Sn, 6%O E. 79% Sn, 21%O
- A compound has the following wt% composition: 2.1% H, 65.3% O, 32.6% S. What is its empirical formula?
 A. H_2SO_4 B. $\text{H}_4\text{S}_2\text{O}_8$ C. S_4HO_2 D. H_2SO_2 E. HO_4S
- How many gram Na in 5.00g Na_2SO_4 ?
 A. 0.035 g B. 1.31 g C. 1.62 g D. 0.97 g E. 0.81 g
- A compound containing only carbon and hydrogen has 85.7 wt% C and 14.3 wt% H. Its molar mass is 56 amu. What is its molecular formula?
 A. C_4H_8 B. C_2H_4 C. C_3H D. CH_2 E. CH
- Balance the following reactions. No fractional coefficients, coefficients must be the lowest whole numbers.
 a) K + H_2O \rightarrow KOH + H_2
 b) H_2O_2 \rightarrow H_2O + O_2
 c) C_2H_6 + O_2 \rightarrow CO_2 + H_2O
 d) KOH + H_3PO_4 \rightarrow K_3PO_4 + H_2O
 e) $\text{C}_2\text{H}_5\text{OH}$ + O_2 \rightarrow CO_2 + H_2O
- How many g of CO_2 are formed from the combustion of 2.9 g of butane, C_4H_{10} , with excess oxygen?
 The balanced reaction is: $2\text{C}_4\text{H}_{10} + 10\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$
 A. 0.26g B. 6.6g C. 8.8g D. 10.5g E. 2.2g
- How many g I_2 will react with 20.4 g Al to form AlI_3 (aluminum iodide)? As always, first balance the reaction! (Hint: this is not a limiting reagent problem)
 A. 144. g B. 288. G C. 192. G D. 72 g E. 576 g
- For the balanced reaction: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$. How many moles of NH_3 will be produced when 0.50 mol N_2 reacts with 1.0 mol H_2 ?
 A. 0.67 mol B. 1.0 mol C. 0.50 mol D. 2.0 mol E. 3.0 mol
- Calculate the mass of CO_2 formed when 2.3 g of ethanol, $\text{C}_2\text{H}_5\text{OH}$ reacts with 3.2 g oxygen O_2 to form CO_2 and H_2O . (You balanced the reaction in 8.e)! This is a limiting reagent problem!
 A. 4.6g B. 2.2g C. 4.4g D. 6.6g E. 2.93g

Chem 101 Multiple Choice Questions

Mary Millhollon, Richard H. Langley



Chem 101 Multiple Choice Questions:

The Hidden Curriculum - Faculty Made Tests in Science Sheila Tobias, 1997 **Cracks in the Ivory Tower** Jason Brennan, Phillip Magness, 2019-04-01 Academics extol high minded ideals such as serving the common good and promoting social justice Universities aim to be centers of learning that find the best and brightest students treat them fairly and equip them with the knowledge they need to lead better lives But as Jason Brennan and Phillip Magness show in *Cracks in the Ivory Tower* American universities fall far short of this ideal At almost every level they find that students professors and administrators are guided by self interest rather than ethical concerns College bureaucratic structures also often incentivize and reward bad behavior while disincentivizing and even punishing good behavior Most students faculty and administrators are out to serve themselves and pass their costs onto others The problems are deep and pervasive most academic marketing and advertising is semi fraudulent To justify their own pay raises and higher budgets administrators hire expensive and unnecessary staff Faculty exploit students for tuition dollars through gen ed requirements Students hardly learn anything and cheating is pervasive At every level academics disguise their pursuit of self interest with high faluting moral language Marshaling an array of data Brennan and Magness expose many of the ethical failings of academia and in turn reshape our understanding of how such high power institutions run their business Everyone knows academia is dysfunctional Brennan and Magness show the problems are worse than anyone realized Academics have only themselves to blame **Chemistry: The Easy Way** Joseph A. Mascetta, Mark Kernion, 2019-08-06 A self teaching guide for students Chemistry *The Easy Way* provides easy to follow lessons with comprehensive review and practice This edition features a brand new design and new content structure with illustrations and practice questions An essential resource for High school and college courses Virtual learning Learning pods Homeschooling Chemistry *The Easy Way* covers Atomic Structure Chemical Formulas Electrochemistry The Basics of Organic Chemistry And more **CliffsAP Chemistry, 4th Edition** Bobrow Test Preparation Services, 2011-09-26 Your complete guide to a higher score on the AP Chemistry exam Why CliffsAP Guides Go with the name you know and trust Get the information you need fast Written by test prep specialists Contents include Introduction overview of the test and how it is scored proven strategies for each type of question Review of topics tested atom periodic table bonding geometry hybridization stoichiometry gases liquids and solids thermodynamics solutions equilibrium acids and bases kinetics redox nuclear chemistry organic chemistry and writing reactions The Labs feature 20 multiple choice questions multiple free response questions on each topic with answers on each topic with answers and explanations scoring rubrics and 2 full length practice exams Structured like the actual exam Complete with answers and explanations AP is a registered trademark of the College Board which was not involved in the production of and does not endorse this product **MEDICAL BIOCHEMISTRY** DR. MANORAMA SWAIN, 2018-01-01 *The Book Medical Biochemistry an Approach to MBBS Examination* is a useful book for 1st Professional MBBS students for revision and last hour preparation The book contains university

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 - Chapter 4: Chem 101 Multiple Choice Questions in Specific Contexts
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