

Chemistry Study Guide

Periodic Table of the Elements

| | | | | | | | | | | | | | | | | | | | | | | | | | | | |
|---|----|----|----|----|----|----|----|----|----|----|----|----|----|-----|----|-----|----|-----|-----|----|----|----|----|----|----|----|----|
| 1 | 2 | | | | | | | | | | | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 1 | H | He | | | | | | | | | | | | | | | | | | | | | | | He | | |
| 2 | Li | Be | | | | | | | | | | | B | C | N | O | F | Ne | | | | | | | | | |
| 3 | Na | Mg | | | | | | | | | | | Al | Si | P | S | Cl | Ar | | | | | | | | | |
| 4 | K | Ca | Sc | Ti | V | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr | | | | | | | | | |
| 5 | Rb | Sr | Y | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In | Sn | Sb | Te | I | Xe | | | | | | | | | |
| 6 | Cs | Ba | | | Hf | Ta | W | Re | Os | Ir | Pt | Au | Hg | Tl | Pb | Bi | Po | At | Rn | | | | | | | | |
| 7 | Fr | Ra | | | Rf | Db | Sg | Bh | Hs | Mt | Ds | Rg | Cn | Uut | Fl | Uup | Lv | Uus | Uuo | | | | | | | | |
| | | | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb | Lu | | | | | | | | | | |
| | | | Ac | Th | Pa | U | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No | Lr | | | | | | | | | | |

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** ✨🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|----------|--------|------------|--------|-------------|
| Proton | p^+ | 1 | +1 | Nucleus |
| Neutron | n | 1 | 0 | Nucleus |
| Electron | e^- | ~ 0 | -1 | Orbit/Shell |

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The American Educational Catalog ,1920 **Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples **The Publishers Weekly** ,1917 **The Publishers' Trade List Annual** ,1992 Chemistry Basics (Speedy Study Guide) Speedy Publishing,2014-12-28 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office,1967 Includes Part 1 Number 1 Books and Pamphlets Including Serials and Contributions to Periodicals January June *Educational Publication* North Carolina. Department of Public Instruction,1927 **Study Guide for Chemical Principles** Thomas Elliott Taylor,1979 **Books and Pamphlets, Including Serials and Contributions to Periodicals** Library of Congress. Copyright Office,1965 *Platers' Guide* ,1916 **Chemical Abstracts** ,1928 Resources in Education ,1976 **Industrial & Engineering Chemistry** ,1928 *Peterson's Annual Guides to Graduate Study* Peterson's Guides, inc,1976 *Machine Drawing* Carl Lars Svensen,1921 Visual illusions, their causes, characteristics and applications Matthew Luckiesh,1922 A Short Course in the Testing of Electrical Machinery for Non-electrical Students John Harold Morecroft,Frederick William Hehre,1921 *The Annual Guides to Graduate Study* ,1973 The Cumulative Book Index ,1909 A world list of books in the English language Competition Science Vision ,2005-10 Competition Science Vision monthly magazine is published by Pratiyogita Darpan Group in India and is one of the best Science monthly magazines available for medical entrance examination students in India Well qualified professionals of Physics Chemistry Zoology and Botany make contributions to this magazine and craft it with focus on providing complete and to the point study material for aspiring candidates The magazine covers General Knowledge Science and Technology news Interviews of toppers of examinations study material of Physics Chemistry Zoology and Botany with model papers reasoning test questions facts quiz contest general awareness and mental ability test in every monthly issue

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