

Chapter 1 – Science and Measurement

Memorize: SI conversions, °C to °F conversion, and density formula (or units)

Matter: mixtures vs pure substances (elements and compounds)

Definition of a solution. Gases combine to make solutions.

Chemical vs physical changes/properties.

Energy: The law of conservation of energy. Energy cannot be created or destroyed, only converted from one form to another.

Know how to convert between SI units.

Mega-	M	10^6
kilo-	k	10^3
deci-	d	10^{-1}
centi-	c	10^{-2}
milli-	m	10^{-3}
micro-	μ	10^{-6}
nano-	n	10^{-9}
pico-	p	10^{-12}

Conversions between SI units and any other units.

Units of area, volume, and density.

Significant figures.

Know how to tell how many sig. figs. a given number has.

Know how to read instruments to the correct sig figs.

Know how to do multi-step calculations to give the answer to correct sig figs.

Know how to round to the correct number of sig figs.

Precision vs Accuracy

Use units in dimensional analysis as a guide in setting up calculations.

Know density units and how to do calculations to yield either mass, volume, or density.

Be able to convert temperatures between °C, °F and K.

Chemistry Mixture And Solution Study Guide Key

SMITH



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Study Guide for Chang's Chemistry Kenneth W. Watkins,1988 **Study Guide/Selected Solutions Manual** Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Fundamentals of Chemistry, Study Guide** James E. Brady,John R. Holum,1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

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