
Chemistry (Semester 2) Final Exam Questions And Answers

(?) according to the kinetic molecular theory, the particles which make up a gas -
CORRECT ANSWER-are in constant random motion

(?) solutions which contain more solute than normal - CORRECT ANSWER-saturation

(?) the maximum amount of solute which can dissolve into a solvent -
CORRECT ANSWER-solubility

(?) the rate at which a gas diffuses is not affected by - CORRECT ANSWER-
the speed of its particles

6.02×10^{23} (to the 23rd) - CORRECT ANSWER-avogadro's number

6.02×10^{23} (to the 23rd) particles - CORRECT ANSWER-mole

a compound that has a specific number of water molecules bound to its atoms -
CORRECT ANSWER-hydrated crystal

a formula used to represent a group of covalently bonded atoms - CORRECT ANSWER-
empirical formula

a homogenous mixture in which one or more substances is dissolved in another -
CORRECT ANSWER-solution

a number written in front of a chemical symbol - CORRECT ANSWER-coefficient

a reaction in which a compound breaks down into two or more elements of new compounds - CORRECT ANSWER-decomposition reaction

a reaction in which two or more substances react to produce a single product -
CORRECT ANSWER-synthesis reaction

a reaction involving the exchange of positive ions between two compounds dissolved in water - CORRECT ANSWER-double replacement reaction

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