

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. direction of electron flow \_\_\_\_\_
- f. direction of ion movement \_\_\_\_\_
- g. half-cell reduction equation \_\_\_\_\_
- h. cell \_\_\_\_\_
- i. Write overall cell equation with phases.  $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$  (do not use arrows, include all reactants/products, and phases)

\_\_\_\_\_

# Voltaic Cells Study Guide

**Donald W. Shive, Louise E. Shive**



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