

Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**

12. List (in the correct order) the steps of the scientific method.

1. make an observation	4. test hypothesis (experiment)
2. ask a question	5. analyze data/form a conclusion
3. form a hypothesis	6. develop a theory

13. What mnemonic device is used to remember the metric system?

K H Da B d c m
g, m, L

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	K	H	Da	g, m, L	d	c	m
Meaning in powers of 10	10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}

Convert the following:

- | | |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL | 21. 25 cm to <u>0.25</u> m |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg | 23. 15 g to <u>0.015</u> kg |
| 19. 0.35 kg to <u>350</u> g | 24. 3.452 kL to <u>3452</u> L |
| 20. 300 mL to <u>0.300</u> L | 25. 47 m to <u>47000</u> mm |

Scientific Notation:

Convert the following to scientific notation:

- | | | | |
|------------|--------------------------------------|------------|--|
| 26) 0.005 | <u>5×10^{-3}</u> | 30) 0.025 | <u>2.5×10^{-2}</u> |
| 27) 5.050 | <u>5.05×10^0</u> | 31) 0.0025 | <u>2.5×10^{-3}</u> |
| 28) 0.0008 | <u>8×10^{-4}</u> | 32) 500 | <u>5×10^2</u> |
| 29) 1,000 | <u>1×10^3</u> | 33) 5,000 | <u>5×10^3</u> |

Chocolate Unit Review Study Guide Answer Key

R Bogdan



Chocolate Unit Review Study Guide Answer Key:

Whispering the Strategies of Language: An Emotional Quest through **Chocolate Unit Review Study Guide Answer Key**

In a digitally-driven world where monitors reign great and quick interaction drowns out the subtleties of language, the profound techniques and psychological subtleties concealed within words frequently move unheard. However, set within the pages of **Chocolate Unit Review Study Guide Answer Key** a fascinating literary treasure pulsating with natural feelings, lies an exceptional journey waiting to be undertaken. Written by a talented wordsmith, that marvelous opus attracts viewers on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within the very material of every word. Within the emotional depths of the touching review, we can embark upon a genuine exploration of the book's core subjects, dissect their interesting writing model, and fail to the strong resonance it evokes deep within the recesses of readers' hearts.

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